A Simple and Rapid Method for the Determination of Nicotine in Third-Hand Smoke by Liquid Chromatography and Its Application for the Assessment of Contaminated Outdoor Communal Areas

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Abstract

This is the first report on the determination of nicotine in third-hand smoke (THS) in outdoor communal areas. THS can be defined as the contamination of surfaces by second-hand smoke. This can remain for periods of time and undergo further chemical reactions to produce further pollutants which can be re-suspended in dust or re-emitted into the gas phase. As THS is a rather complex mixture studies have focused on using nicotine as a marker of THS, as it is the most abundant organic compound emitted during smoking. In this present study, the extraction of dust wipe samples and the subsequent chromatographic conditions required for the separation of nicotine by hydrophilic interaction liquid chromatography were optimised. The optimum chromatographic conditions were identified as a 150 mm x 4.6 mm, 5 μ m Zorbax Carbohydrate Analysis column with a mobile phase consisting of 90 % acetonitrile, 10 % water at a flow rate of 1.0 mL/min with UV detection at 259 nm. Further investigations were made on samples collected from surfaces of public entrance ways. Under these conditions, a linear range for nicotine of 0.05 to 24 μ g/mL (1.0 ng – 480 ng on column) was obtained, with a detection limit of 1.0 ng on column based on a signal-to-noise

ratio of three. Acetone, naphthalene, phenol, musk ketone and palmitic acid were found not to interfere. Entrances were found to be contaminated with THS nicotine between 5.09 μ g/m² and 309 μ g/m² notably higher than that found in studies of indoor environments.

Keywords: Third-hand smoke; HILIC; dust wipe; nicotine; communal areas; outdoor

Introduction

The terms first and second-hand smoke have been used for some time and commonly first-hand smoke is defined as what is inhaled into the lungs of the smoker, while second-hand smoke is a mixture of exhaled smoke and other substances leaving the smouldering end of the cigarette that enters the atmosphere and can be inhaled by others. The term third-hand smoke (THS) is relatively new [1-3], defined as the contamination that remains on the surfaces after smoking source has been removed [4,5]. Nicotine present in the tobacco smoke residue can combine with other compounds such as ozone and nitrous oxide to produce recognized carcinogens such as nitrosamines [6-11]. Studies by Martins-Green *et al* [6] have shown that mice exposed to THS exhibit poor healing processes, produce excess collagen and high levels of inflammatory cytokines, conditions which can lead to inflammation-induced diseases such as chronic obstructive pulmonary disease and asthma. Behavioural tests undertaken in the same investigation showed THS-exposed mice also exhibited increased hyperactivity [6]. The effects of THS are predicted to be more severe for children as they are still developing and also ingest more dust particles as they explore objects with their mouths [12-15].

In the UK, smoking has been banned in enclosed buildings and indoor areas since the 1st July 2007 as a consequence of the Health Act 2006. This has resulted in people generally congregating in areas close to building entrances in order to smoke. This can result in both issues with second-hand smoke and third-hand exposure to themselves and unrelated people in the vicinity and when using entrance ways.

As THS is a rather complex mixture, studies have focused on using nicotine as a marker as it is the most abundant organic compound emitted during smoking [16] and a number of different methods

have been reported for its determination [17], such as ion mobility spectroscopy [18] and voltammetry [19]. Previous studies of THS have utilised sophisticated and expensive analytical equipment such as tandem mass spectroscopy and two-dimensional gas chromatography (table 1). The analysis of some amines such as nicotine by reverse phase HPLC can lead to problems due to their hydrophilic nature leading to low retention; necessitating the use of ion pairing agents [20], buffers, or low concentrations of organic modifier [21] which can cause stationary phase de-wetting. The presence of ionised silanol groups on the stationary phase can also result in peak tailing and overloading leading to poor separation and analytical performance [22,23]. Hydrophilic interaction liquid chromatography (HILIC) is becoming an established alternative technique to overcome these issues [24,25]. The high percentages of acetonitrile generally used with HILIC result in mobile phases with low viscosities allowing for greater flow rates ranges to be utilised resulting in increased analytical throughput and low backpressures which can lead to improved signal-to-noise ratios. In this present study we have used the collection of dust wipe samples to investigate the extent of THS within our university campus to assess exposure to individuals. Dust wipe samples were analysed by HILIC with UV detection to identify and quantify nicotine present. This represents a simpler and cheaper alternative method using instrumentation normally found in most laboratories.

Analytical Technique	Nicotine Levels Third-Hand Smoke (µg/m ²)	Comments	Ref.
Liquid	Hotels with complete smoking ban guestrooms, mean 1.4,	Air concentrations of nicotine also investigated. Air and	[26]
chromatography-	hallways, 1.2.	surface concentrations of 3-ethynylpyridine	
tandem mass	Hotels without complete smoking ban; non-smoking	concentrations also investigated.	
spectrometry	guestrooms, mean 3.7; smoking guestrooms, 51.8. Hallways		
	outside non-smoking rooms, 2.8; hallways outside smoking		
	rooms, 9.3.		
Liquid	Non-smoking environment 1.8 – 5.3. Smoking environments	Wipe sampling for nicotine in private homes, private cars,	[10]
chromatography-	57.2 – 113.5.	rental cars, and hotels with and without smoking bans.	
tandem mass			
spectrometry			
Liquid chromatography	Geometric mean surface nicotine concentrations of non-		[27]
tandem mass	smoking and smoking homes 11.4, 90.9 respectively between		
spectrometry	homes with complete, partial and no voluntary home smoking		
	restrictions 8.9, 56.3 vs. 145.6 respectively.		
Liquid	Living rooms; geometric mean, 83.67. Child bedrooms:	Analysis of the homes of Hookah smokers. Indoor air	[28]
chromatography-	geometric mean, 64.6.	nicotine, 4-(methylnitrosamino)-1-(3-pyridyl)-1-butanone	
tandem mass		(NNK), acrolein, cotinine, 4-(methylnitrosamino)-1-(3-	
spectrometry		pyridyl)-1-butanol (NNAL) and NNAL-glucuronides (total	
		NNAL) and 3-hydroxypropylmercapturic acid also	
		determined.	
Liquid		Surface dust samples from both the homes of smokers	[29]
chromatography-		and non-smokers for nicotine derived carcinogen, 4-	
tandem mass		(methylnitrosamino)-1-(3-pyridyl)-1-butanone (NNK).	
spectrometry		Non-smokers' properties; <30 pg/100 cm ² – 435 pg/100	
		cm ² ; smokers'; 37 pg/100 cm ² – 3500 pg/100 cm ² .	
Liquid	Laboratory study.	Cotton terry cloth and polyester fleece were exposed to	[30]
chromatography-		smoke in controlled laboratory conditions and aged	
tandem mass		before extraction. Nicotine, myosmine, 2,3-bipyridine,	
spectrometry		cotinine, N-formylnornicotine, nicotelline (3,2':4',3''-	
		terpyridine) NNN, NNK, NNA determined. Aqueous and	
		solvent (methanol/HCl) efficiency compared.	
Hydrophilic interaction	Surface nicotine non-smoker living rooms greater or equal to	Urine cotinine, finger and air nicotine measurements	[31]

liquid chromatography- tandem mass spectrometry	5.0, 16 %; smoker homes 86 %; surface nicotine non-smoker bedrooms greater or equal to 5.0, none detected; smoker homes 75 %. Dust nicotine non-smoker living rooms greater or equal to 5.0, 28 %; smoker homes 90 %; dust nicotine non- smoker bedrooms greater or equal to 5.0, none detected; smoker homes 84 %; dust nicotine non-smoker living rooms greater or equal to 5.0, 31 %; smoker homes, 91 %.	made as well.	
Gas chromatography with nitrogen selective detection	Experimental investigation using model materials (window, walls, floor, wood, and metal). Three brands of e-cigarettes were refilled with varying nicotine concentrations. One hundred puffs were released from each product directly into an exposure chamber. Surface wipe samples were taken from 5 indoor 100 cm ² surfaces and extracted with 5 mL of methanol.	Three of the four experiments showed significant increases in the amount of nicotine on all five surfaces. The floor and glass windows had the greatest increases in nicotine, on average by a factor of 47 and 6, respectively.	[32]
Gas chromatography with nitrogen selective detection and liquid chromatography- electrospray ionization-tandem mass spectrometry	No evidence for third-hand smoke exposure in this study.	Children's urine examined for nicotine, cotinine and 4- (methylnitrosamino)-1-(3-pyridyl)-1-butanol (NNAL).	[33]
Gas chromatography/mass spectrometry	Furniture 11 – 73; household dust 0.89 – 4.43; vehicle dashboard 5.0 – 8.6; vehicle dust 11.6 – 19.5.	Air nicotine and finger wipes also investigated.	[34]
Comprehensive gas chromatography nitrogen chemiluminiscence detection	Smokers' house dust 4.33 – 342. Non-smokers' house dust 0.62 - 5.3.	Main tobacco specific <i>N</i> -nitrosamines also determined.	[8]
Gas chromatography– mass spectrometry	Levels ranging from 0.184 – 0.256 for households with one or more non-smoking members; 0.415 – 1.634 households with smokers.	House dust was collected by high-volume surface samplers and from household vacuum cleaners and analysed for nicotine.	[35]

Gas chromatography-	<0.020 - 7.776	Household carpet dust. Polychlorinated biphenyls and	[36]
mass spectrometry		polyaromatic hydrogens also measured.	
Gas chromatograph	Experimental investigation using model materials (cotton,	Along with nicotine, ozone oxidation products cotinine,	[37]
ion trap – tandem	paper, and gypsum wallboard) exposed to ten smouldering	myosmine and N-methyl formamide and nicotine-1-oxide	
mass spectrometry	cigarettes in a 24 m ³ room size chamber in the presence and	detected.	
	absence of ozone.		
HILIC UV detection	<0.01 - 309	Contamination of outside doors of communal entrances	This
		by tobacco smoke.	study

Table 1. Determination of third-hand smoke.

Experimental

Chemical and Reagents

Nicotine was obtained from Sigma-Aldrich. The surface-wipes were fabricated by cutting 10 cm² squares from a roll of tissue (one ply, Jangro White Centrefeed, Pattersons, Bristol, UK). Deionised water was obtained from a Purite RO200–Stillplus HP System, (Purite Oxon, UK). Acetonitrile was obtained from Fisher (Loughborough, UK).

Instrumentation

HILIC separation was performed using a system consisting of an IsoChrom pump (Spectra Physics), with a 150 mm x 4.6 mm, 5 µm Zorbax Carbohydrate Analysis column connected to a 7125 valve manual injector fitted with a 20 µL sample loop (Rheodyne, Cotati, CA). Sample extracts were analysed using a mobile phase consisting of 90 % acetonitrile (Fischer, far UV, HPLC grade), 10 % water, at a flow rate of 1.0 mL/min with UV detection at 259 nm. Chromatograms were recorded using a Siemens Kompensograph X-T C1012 chart recorder. Sample extracts and standards were introduced using full loop injection. Quantification was achieved by manual integration using external calibration.

Dust Wipe Sampling

The glass fronting of doors and windows facing onto entranceways were chosen for investigation. Dust wipe samples were obtained by wiping from the upper left corner of the sample area; in "S" shape manner, wiping from side-to-side whilst moving down the sample area. The area sampled was measured in order that a comparison of concentration values (μ g/m²) between different surfaces could be made. The exposed wipe was then folded in half, exposed side to exposed side and another "S" shape was made in the opposite direction wiping up and down instead of side-to-side. The folded wipe was placed in an acetonitrile washed glass vial, which was also used as the extraction vessel and sealed. A new pair of gloves was also used for each sample. A procedural blank was obtained by taking a tissue on site but without sampling the surface.

Sample Extraction and Analysis

Five mL of acetonitrile was added to the glass vessel containing the dust wipe sample. This was then sealed and the dust wipe extracted by sonication for 15 minutes at room temperature. A 20 μ L aliquot of the resulting solution was then examined using the optimised HILIC conditions.

Results and Discussion

Effect of Acetonitrile/Water Concentration on Chromatographic Performance

The variation of chromatographic capacity factor (k') of nicotine with the percentage of water in the mobile phase is shown in Figure 1. The optimum percentage of water was found to be 10 % (v/v), as this gave a retention time of only 5.5 minutes (k' = 3.2) without compromising the chromatographic performance. Therefore, a mobile phase of 90 % acetonitrile, 10 % water was used in further studies.

Calibration Curve, Limit of Detection, and Precision

Standard solutions containing nicotine in the concentration range 0.05 to 24 μ g/mL (1.0 ng – 480 ng on column) were prepared in acetonitrile and determined by the optimized HILIC procedure. The calibration plot was found to be linear over the range investigated, with an R² value of 0.998. The coefficient of variation was determined by performing six replicate measurements of a 1.50 μ g/mL (30.0 ng on column) nicotine standard and was calculated to be 1.7 %. The limit of detection was calculated by making replicate measurements at the retention time of nicotine, 5.5 minutes (n = 5) for a blank solution. The detection limit based on three times the mean of these measurements gave a value of 1.0 ng of nicotine on column.

Studies of Possible Interferences

A wide range of compounds are known to be present in cigarette smoke [38,39] and potentially present in third-hand smoke [4] which could interfere with the determination of nicotine. Present as well, are residues from cleaning products and human contact. In the present study we have investigated acetone, toluene, phenol, naphthalene, musk ketone and palmitic acid all at 20 μ g/mL as possible interferences. All the compounds investigated were found to elute as part of the unretained fraction and consequently did not interfere.

Extraction and Recovery of Nicotine from Dust Wipes

Optimisation of Extraction Procedure

To assess the performance of the developed method, five replicate determinations of nicotine in spiked and unspiked dust wipe samples were undertaken. Five separate dust wipes were spiked with 5.10 µg of nicotine and extracted with 5 mL of acetonitrile with aid of sonication for 15 minutes as described above in the Sample Extraction and Analysis section. A 20 µL aliquot of the resulting solution was then introduced to the HPLC and quantification was achieved by external calibration. Table 2 shows the summaries the recovery and precision data obtained. A mean recovery of 96.2 % with an associated coefficient of variation of 1.14 % was obtained. As can be seen from this data, the developed method holds promise for the determination of such levels of nicotine.

Sample	Nicotine added, µg	Nicotine found, μg	% Recovery
1	5.10	4.84	94.9
2	5.10	4.95	97.1
3	5.10	4.90	96.1
4	5.10	4.88	95.7
5	5.10	4.96	97.3

Mean recovery = 96.2 %, coefficient of variation = 1.14

Table 2. Recovery and precision data for nicotine obtained for dust wipe samples.

Several reports have shown that poor recoveries can result due to the volatile nature of nicotine [21,40]. This can have dramatic effects in common extraction and concentration steps such as if the sample extract is subject to a nitrogen blow down step. To overcome this problem nicotine is commonly converted to its salt by the addition of acid to the sample extract. To investigate this effect, six separate 5 mL aliquots of acetonitrile were fortified with a 1.0 μ g of nicotine. Three of these were blown down to dryness under nitrogen at room temperature. The remaining three were first acidified with 10 μ L of concentrated HCl and then blown down to dryness with nitrogen. The

resulting extracts were then reconstituted in 1.0 mL of mobile phase and examined by the optimised liquid chromatographic conditions. A mean recovery of 12.1 % and 98.2 % was obtained for the non-acidified and the acidified samples respectively. Consequently, in further studies all samples requiring nitrogen blown down were first acidified.

Extraction and Analysis of Real Samples

Dust wipe samples were collected from the glass panels of external entrance doors and extracted with 5 mL of acetonitrile by sonication as described above. All sample sites were found to be contaminated with nicotine levels that could be directly determine without the need for concentration step and hence did not require blowing down under nitrogen. Table 3 details the nicotine levels obtained for these samples.

Sample	Nicotine, μg/m ²
Entrance 1	5.09
Entrance 2	36.2
Entrance 3	244
Entrance 4	8.10
Smoking shelter	309

Table 3. Nicotine levels obtained for dust wipe samples.

Conclusions

This is the first report to our knowledge of levels of THS nicotine in outdoor communal areas, and is also the first report on the determination of nicotine in dust wipe samples by HILIC with UV detection. This has been shown to be a potentially simpler and economic alternative to commonly employed techniques such as liquid and gas chromatography/mass spectroscopy. A number of the samples investigated had notable concentrations of nicotine which are notably higher than that reported indoor household dust samples [8]. Results indicate that there is a potential issue with exposure to THS and further investigations are required.

Extractions of nicotine can be hampered by the volatile nature of nicotine, leading to poor recoveries resulting from losses during extraction and concentration steps, such as nitrogen blow down [21,41]. For the levels detected in this study a further concentration step was found unnecessary. The developed method avoids the need for such a concentration step and hence avoids such issues.

The introduction of the electronic cigarette has added further issues which have yet to be fully assessed [32,42,43] and further investigations into their effects on THS needs to be addressed. In future studies we will also investigate the possible application of the assay for the determination of other pollutants and drugs [44,45]. HILIC has been shown to be fully compatible with electrochemical detection [25] and to be a sensitive method for the determination of nicotine [21] we will also investigate this as a possible alternative detection system.



Figure 1. Variation of capacity factor (k') with percentage of water in mobile phase.



Figure 2. Representative chromatograms obtained for THS dust wipe samples for (a) Entrance 2; (b) Entrance 1; (c) Entrance 3 and (d) procedural blank. * = nicotine

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