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Highlights

Chemical composition, antioxidant activity and sensory evaluation of five different species of brown edible seaweeds

Food Research International xxx (2014) xxx - xxx

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- ^c Centre for Research in Biosciences, The University of the West of England, Frenchay Campus, Bristol BS16 1QY, UK
- The composition and sensory profile of five seaweeds was evaluated.
- Fucus sp. and Ascophyllum nodosum showed high antioxidant activities.
- Nucleotide in Fucus vesiculosus was 10 times higher than reported in other foods.
- Laminaria was significantly different according to panellists.

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Chemical composition, antioxidant activity and sensory evaluation of five different species of brown edible seaweeds

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- 40

ABSTRACT

The chemical and volatile composition as well as sensory profile of five brown edible seaweeds collected in the 30 United Kingdom, was evaluated. The ash content was 190-280 mg/g, NaCl 35.1-115.1 mg/g, protein 2.9-6.0 g/g, 31 and fat 0.6-5.8 g/g (dry basis). Fucus vesiculosus, Fucus spiralis and Ascophyllum nodosum showed higher antiox- 32 idant activities (DPPH and FRAP). Nucleotide concentrations were of the same order of magnitude as reported in 33 other foods such as tomatoes or potatoes, except for F. vesiculosus where levels of nucleotides were 10 times 34 higher. The fatty acids profile was dominated by oleic acid (21.9–41.45%), followed by myristic (6.63–26.75%) 35 and palmitic (9.23-16.91%). Glutamic and aspartic acids (0.15-1.8 mg/g and 0.05-3.1 mg/g) were the most 36 abundant amino acids. Finally, sensory and volatile analyses illustrated that Laminaria sp. had the strongest sea- 37 weed and seafood-like aroma and taste.

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1. Introduction

Due to their low content of lipid, high concentration of polysaccharides, natural richness in minerals, polyunsaturated fatty acids and vitamins as well as their high content of bioactive molecules, marine algae have, in recent years, received great attention (Gupta & Abu-Ghannam, 2011a,b). Algae are grouped into two main categories; the microalgae, found in both benthic and littoral habitats and also throughout the ocean waters as phytoplankton, and the macroalgae or seaweeds, which occupy the littoral zone, and can be classified as red (Rhodophyta), brown (Phaeophyta) or green (Chlorophyta), depending on their nutrient and chemical composition (Dawczynski, Schubert, & Jahreis, 2007; Gupta & Abu-Ghannam, 2011a).

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Red and brown algae are mainly used, within the traditional Japanese 56 diet as sushi wrappings, seasonings, condiments and vegetables and can 57 thus constitute between 10% and 25% of food intake of most Japanese 58 people. Although the principal uses of seaweeds in Europe are as a 59 source of phycocolloids (thickening and gelling agents) for various 60 industrial applications, including uses in foods or as feed and fertiliser 61 (Ortiz, Bozzo, Navarrete, Osorio, & Rios, 2006; Yaich et al., 2011), con-62 sumption of seaweed products has recently increased with currently, 63 approximately 15-20 edible algae species being commonly marketed 64 for consumption. These seaweed varieties differ greatly in their quality, 65 colour, consistency, and nutrient content (Dawczynski et al., 2007; 66 Mišurcová, 2011; Mišurcová, Ambrožová, et al., 2011; Mišurcová, 67 Machů, et al., 2011). Different authors have pointed out that the chem- 68 ical composition of seaweeds varies with species, habitats, maturity and 69 environmental conditions (Ortiz et al., 2006; Sanchez-Machado, 70 Lopez-Cervantes, & Lopez-Hernandez, 2004).

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The European seaweed industry is dominated by Norwegian, French and Irish production, while Spain, Portugal and the UK are small producers and suppliers. Particularly, in the UK, the market for seaweed (therapeutic, biotechnology, bio-fuel seaweeds based, or foods) is mostly imported, whereas there is abundance of growing seaweeds around the islands, with some local producers already harvesting them for commercial purposes. Particularly, in the coast of Scotland there are dozens of different kinds of edible seaweed, being the red seaweed dulse (*Palmaria palmata*), as well as the brown seaweeds: kelp (*Laminaria sp.*) and different wracks (*Fucus sp.*, *Ascophyllum nodosum*, *Pelvetia canaliculata*) the most generally harvested (due to their abundance and accessibility).

The use of brown seaweeds, as ingredient or as a whole food, has already been reported by numerous authors to be beneficial in different aspects. For instance, as an alternative source of protein, with some brown species having higher protein content than soybeans. Their fat content accounts for 1 to 6 g/100 g dry weight with some varieties, as Laminaria sp. generally between 1.5 and 3.3% of dry weight (Fleurence, Gutbier, Mabeau, & Leray, 1994), and some of these species are also characterised by a high level of eicosapentaenoic acid (up to 24% of the total fatty acid fraction) (Fleurence, 2004). Antioxidants are also other important metabolites in brown seaweeds including fucoxanthin, polyphloroglucinol, phenolic compounds or bromophenols, that have been isolated from species such as *Fucus* and *Laminaria* (Fleurence et al., 2012; Gupta & Abu-Ghannam, 2011b; Xu, Fan, et al., 2004; Xu, Song, et al., 2004).

In addition, there are recent projections in the functional effects of seaweeds as means to improve the fibre content and reduce the salt content of food products. This is mainly due to their high content in umami compounds such as nucleotides or some amino acids. The aim of this study was to characterise five different brown edible seaweeds locally produced on the west coast of Scotland (Isle of Bute), UK, in terms of chemical composition as well as sensory and volatile analyses; this information might be useful to evaluate their use as food ingredients and their potential contribution to the diet.

2. Material and methods

2.1. Raw material

Five different species of brown seaweed (*Laminaria digitata*, *Ascophyllum nodosum*, *P. canaliculata*, *Fucus vesiculosus*, *and Fucus spiralis*), were obtained from the same supplier and harvested between May and August 2012 in the west coast of Scotland, United Kingdom. The samples were then freeze-dried and separated into two different batches depending on the harvesting time; seaweeds collected in May and June (batch 1), and those collected in July and August (batch 2). Samples were milled in a mechanical grinder for 10 min, to obtain a fine and homogeneous powder before performing the analyses.

2.2. Chemical analyses

All the chemical analyses were carried out in triplicate on the homogeneous powder.

2.2.1. Dry matter, ash and NaCl content

The dry matter, ash and sodium chloride content were ascertained according to the Association of Official Analytical Chemists (AOAC, 2000).

2.2.2. Protein

Total protein was determined by the Kjeldahl method. The protein was calculated using a nitrogen conversion factor of 6.25 (Ortiz et al., 2006; Yaich et al., 2011). Data were expressed as percentage of dry weight.

2.2.3. Extractable fat

The extractable fat was determined using the Soxhlet extraction 131 method with petroleum ether 40:60 as solvent. (AOAC, 2000). 132

2.2.4. Fatty acids

The fatty acid composition was analysed by GC-FID after transes- 134 terification to methyl esters (FAMEs) with a mixture BF₃ methanol at 135 20 °C according to the IUPAC standard method (IUPAC, 1992; Yaich et al., 2011).

Fat (10 mg), hexane (0.2 mL) and BF₃ (0.5 mL) were heated at 70 °C 138 for 1.5 h. After transesterification, saturated salt solution (0.5 mL, 25% 139 NaCl), H₂SO₄ (0.2 mL, 10%) and hexane (7 mL) were added to the reaction medium. Analysis of FAMEs was carried out with a Hewlett Packard 141 6890 GC equipped with an auto sampler, an Agilent 6890 Network FID 142 and an Agilent DB-23 (60 m × 0.25 mm, 0.25 µm) capillary column. The 143 oven temperature was programmed from 90 °C to 240 °C at 4 °C/min 144 and the injector and detector temperatures were set at 250 °C. The 145 carrier gas was helium at 1.0 mL/min constant flow (split ratio 10:1). 146 The software used for data acquisition and processing is 6890N. Data 147 analysis identification and quantification of FAMEs was accomplished 148 by comparing the retention times of the peaks with those of pure standards (Supelco® 37 Component FAME Mix, Sigma) and analysed under 150 the same conditions. The results were expressed as percentage of individual fatty acids in the lipid fraction.

2.2.5. Antioxidants

Seaweed powder (0.1 g) was mixed with 2.5 mL ethanol (95%), 154 vortexed for 30 s and stored at $-20\,^{\circ}\mathrm{C}$ overnight. The sample was centrifuged for 10 min at 2000 \times g at room temperature under dark conditions and the supernatant was used for analysis.

The radical scavenging activity (DPPH), was determined following 158 the modified protocol of Brand-Williams, Cuvelier, and Berset (1995). 159 Sample (10 μ L) and deionized H₂O (90 μ L) were added in a 96-well microtiter plate and the reaction started by adding 200 μ L of freshly prepared DPPH solution (0.024 g/L DPPH). The absorbance was measured 162 at 515 nm every 4 min for 32 min in total, when the absorbance value 163 remained constant.

The reducing power of the samples (FRAP), was determined by the 165 modified protocol described by Benzie & Szeto (1999) and Bub et al. 166 (2000), in a 96-well microtiter plate, following a similar procedure as 167 for DPPH. In this case the reaction was started by adding pre-warmed 168 FRAP reagent (200 µL, 37 °C), the absorbance was determined at a 169 wavelength of 593 nm and the reaction time was 8 min at 37 °C.

Finally, the total phenolic content (TPC) was determined follow- 171 ing the modified protocol of the microplate Folin–Ciocalteu assay 172 (Magalhães, Santos, Segundo, Reis, & Lima, 2010). Samples (50 μ L, 173 [1:10 v/v]) were added to Na₂CO₃ solution (100 μ L, 6% [w/v]). The re- 174 action was started by adding the Folin–Ciocalteu solution (50 μ L, 175 [1:25 v/v]), and the absorbance determined at 725 nm every 5 min 176 for a total of 30 min, when the absorbance value remained constant. 177

For the DPPH and FRAP assay calibration curves of Trolox 178 (0–1000 mM) were prepared and results were expressed as the number 179 of equivalents of Trolox (mmol eq of Trolox/g dry weight). Gallic acid 180 (0–1000 mM) was used for TPC and results expressed as the number 181 of equivalents of gallic acid (mmol eq of gallic acid/g dry weight of sea- 182 weed powder).

2.2.6. Nucleotides

Nucleotides were extracted using water and hydrochloric acid 185 following centrifugation based on a modified version of the protocol 186 by Oruña-Concha, Methven, Blumenthal, Young, and Mottram (2007). 187 Freeze-dried samples (0.3 g) were weighed into 15 mL screw-top 188 vials; distilled water (5 mL) and hydrochloric acid (5 mL, 0.01 N, HCl) 189 were added followed by stirring at 90 °C for 90 min. The mixture was 190 allowed to stand for another 20 min and aliquots of the supernatant 191 (1.5 mL) were centrifuged at 8500 $\times g$ for 15 min.

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The 5′_-nucleotides were separated using a Dionex Ultimate 3000 HPLC system attached to a UV-spectrophotometric detector, HPG-3200 pump, and a 10 μ L sample loop, using solvent A (KH₂PO₄ 0.04 M, pH 5.5) and solvent B (KH₂PO₄ 0.5 M, pH 5.5) as a mobile phase. Gradient elution was carried out as follows: 0–15 min 100% A, 15–20 min 100% B, 20–25 min 100% A (initial conditions), 25 minute re-equilibration wash with 100% A, at a flow rate of 1 mL/min, using a SphereClone 5 μ m SAX 80 Å, LC Column 250 × 4.6 mm (Phenomenex [phenomenex.com]), and UV detection at 254 nm. Each 5′_-nucleotide was quantified using a calibration curve of the pure 5′_-nucleotide (5′_-guanosine monophosphate (GMP), 5′_-inosine monophosphate (IMP) 5-adenosine monophosphate (AMP) and uridine monophosphate, (UMP)). Recovery rates were determined by standard addition methodology.

2.2.7. Amino acids

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An aliquot of the extract used for nucleotides analyses (100 µL) was derivatised using the EZ-Faast amino acid kit (Phenomenex, Torrance, CA). GC-MS analysis were carried out using an 6890 GC coupled to a 5973 MSD instrument (Agilent, Palo Alto, CA) as described by Elmore, Koutsidis, Dodson, Mottram, & Wedzicha (2005). Norvaline was used as internal standard and calibration curves were used for the quantification of the amino acids.

2.2.8. Volatiles analysis

GC-MS analysis was performed using an Agilent 7890A gas chromatograph equipped with a CPWAX capillary column (60 m \times 0.25 mm i.d. \times 0.25 µm FT) and coupled to a BenchToF Time of Flight Mass Spectrometer (Almsco, UK) and a CTC CombiPal autosampler (CTC Analytics AG, Zwingen, Switzerland). HS-SPME was performed on the aqueous extracts used for sensory evaluation (200 µL) in 2 mL of saturated NaCl solution. The samples were incubated at 40 °C for 40 min followed by a 1 min extraction using a CAR/PDMS/DVB SPME fibre and desorption at 260 °C for 10 min. The oven temperature was programmed as follows: initial temperature 40 °C (held for 5 min), 40–200 °C at 4 °C/min, then to 250 °C at 8 °C/min, held for 5 min. Helium was used as the carrier gas at a flow rate of 1 mL/min.

The volatile compounds were identified by comparing their mass spectra (m/z values of the most important ions) with spectral data from the National Institute of Standards and Technology 2002 library as well as retention indices published in the literature (pherobase. com). Relative retention indices were determined by injection into the column of a solution containing the homogenous series of normal alkanes (C_7 – C_{30} ; by Sigma-Aldrich) in the same temperature programmed run, as described above. Quantification of selected compounds was carried out using external calibration curves.

2.2.9. Sensory evaluation

Aqueous extracts in mineral water (1%, w/w) were heated at 70 °C for 30 min and filtered before sensory evaluation. This temperature was chosen as the enzymic degradation processes which break down the RNA into 5'-nucleotides are pH and temperature dependant; and as temperature increases during heating of the samples, nuclease activity increases to around 65–75 °C (Solms & Wyler, 1979; Yang, Lin, & Mau, 2001). Extracts were analysed by conventional sensory profiling, using a non-trained panel (n = 21; 9 female, 12 male). The size of the panel used could be considered small for the general requirements of a conventional sensory profile; nevertheless, for the aim of this sensory study, which was to get a general idea of the perception of the attributes by consumers that would not be very familiar with that kind of product, the use of that sort of panel would be enough according to some previous studies (Clapperton & Piggott, 1979; Delahunty, Mc Cord, O'Neille, & Morissey, 1997; Husson, Le Dien, & Pagés, 2001; Husson' & Pagés, 2003). The sensory attributes studied, which had been previously described by 4 assessors, were: honey-like odour, herbal odour, seaweed-like odour, seafood-like taste, saltiness, astringency, bitterness, green tea-like taste, and salmon-like taste. 10 mL of each seaweed extract at room temper- 256 ature was served to each panellist. Continuous non-structured scales 257 were used for evaluation. The left side of the scale corresponded to the lowest intensity (value 0) and the right side to the highest intensity 259 (value 10). Each panellist rinsed their mouth with mineral water and 260 ate a piece of plain cracker between samples.

2.3. Statistics 262

Analysis of variance (ANOVA) and the Friedman test (p-value < 0.05) 263 were carried out using SPSS to estimate the differences in composition of 264 the seaweed varieties investigated in this study.

Principal Component Analysis, PCA, (SPSS) was also applied to dif- 266 ferentiate the varieties of seaweeds based on their chemical composition and volatile compound profile. 268

3. Results and discussion

3.1. Dry weight, contents of ash, NaCl, protein and extractable fat

Table 1 illustrates the chemical composition of the five different 271 varieties of seaweed depending on the time of harvest. Significant 272 differences (p < 0.05) were found in their composition depending on 273 season (batch) and also on the species. In general terms, the values ob- 274 tained were of the same order of magnitude as those reported by other 275 authors for brown seaweeds (Gómez-Ordóñez, Jiménez-Escrig, & 276 Rupérez, 2010; Ito & Kanji, 1989; Ortiz et al., 2006; Rioux, Turgeon, & 277 Beaulieu, 2009). It is important to point out the high salt levels (NaCl) 278 presented by F. spiralis and L. digitata. No inter-species or inter-batch 279 differences were found in the protein content for these two seaweeds, 280 their values being similar to those reported by Yaich et al. (2011) 281 (8.46% dry weight) and Ortiz et al. (2006); (10% dry weight), but slightly 282 lower than those reported by other authors for brown seaweeds 283 (Gómez-Ordóñez et al., 2010; Rioux et al., 2009). These differences 284 might be expected as variations in the protein content of seaweeds can 285 be attributed to species differences and seasonal effects (Fleurence, 286 1999; Yaich et al., 2011). Extractable lipid varied among the different 287 species, but was of the same order of magnitude as the contents reported 288 by other authors, such as Ito and Kanji (1989) (0.1–4.9% dry weight) or 289 Gómez-Ordóñez et al. (2010) (0.94–5.97% dry weight). F. vesiculosus and 290 P. canaliculata where the two species with the highest extractable fat 291 content, Differences observed, between batches or species, could be at- 292 tributed to factors such as climate, geographical origin of the seaweed 293 and the method used to extract oil. 294

3.2. Antioxidant activity

The antioxidant activity of the ethanolic extracts of the seaweed samples was analysed by two different methods to accurately reflect 297 all the antioxidants in the samples (Table 1). The FRAP reagent can 298 react with iron (II) and thiol groups (Benzie & Szeto, 1999), while 299 DPPH is expected to react with organic radicals (Chandrasekar, 300 Madhusudhana, Ramakrishna, & Diwan, 2006). The values for the total 301 phenolic content are also presented in Table 1 (mmol equivalents of 302 gallic acid/g dry weight). The estimation of the antioxidant potential using different methods enables a better understanding of the mechanism(s) of antioxidative action of the seaweed extracts.

There were differences between the seaweeds species in terms of 306 their antioxidant activity values with *Fucus sp.* and *A. nodosum* being 307 the ones with the highest values (40–50 mmol Trolox/g dry weight 308 [DPPH], 21–55 mmol Trolox/g dry weight [FRAP]). These values are in 309 the same order of magnitude that those reported previously (Díaz-Rubio, Pérez-Jiménez, & Saura-Calixto, 2009). *Fucus sp.* and *Aschophylum* 311 *sp.* were also found to be the species with the highest antioxidant values among different brown seaweed species by Wang, Jónsdóttir, and 313 Ólafsdóttir (2009). In general terms, DPPH and FRAP values followed

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Table 1

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t1.3

Composition of the seaweed samples: moisture (x^w%), ash (% dry weight), NaCl (mg/g dry weight), protein (g/g dry weight) and fat content (g/g dry weight), antioxidant activity (DPPH and FRAP mET/100 g of dry weight), total phenolic content (mEG/100 g of dry weight), fatty acids composition (g/100 g of total fat), and homogeneous groups obtained from the statistical analysis for the different species of seaweeds and the different batches used (n = 3).

Q2 Q3 Q4	4		Batch	Laminaria digitata	Ascophyllum nodosum	Pelvetia canaliculata	Fucus vesiculosus	Fucus spiralis
Q5 Q6	Fresh	x ^w	1	81.0 ± 0.5	69.0 ± 0.2	64.6 ± 3.2	60.0 ± 0.5	76.7 ± 0.5
			2	81.0 ± 0.5	68.1 ± 2.3	66.4 ± 5.4	58.2 ± 3.0	74.3 ± 0.6
t1.7	Freeze dried	Ash	1	$21.0 \pm 0.2 (a)$	$19.0 \pm 0.2 (a)$	$21.0 \pm 0.2 (a)$	$21.0 \pm 0.2 (a)$	$25.0 \pm 0.2 (c)$
			2	$28.0 \pm 0.2 (d)$	$22.0 \pm 0.2 (b)$	$22.0 \pm 0.2 (b)$	$19.0 \pm 0.2 (a)$	$26.5 \pm 0.7 (c)$
		NaCl	1	$91.7 \pm 1.0 (c)$	41.8 ± 0.2 (b)	$35.1 \pm 0.6 (a)$	51.2 ± 0.3 (b)	$94.6 \pm 1.7 (c)$
			2	$115.1 \pm 0.2 (d)$	$61.1 \pm 0.4 (b)$	$51.3 \pm 0.7 (b)$	$49.8 \pm 4.0 (b)$	$93.1 \pm 4.3 (c)$
		Protein	1	5.79 ± 0.08 (b)	5.24 ± 0.01 (b)	$7.26 \pm 0.30 (c)$	5.80 ± 0.17 (b)	5.89 ± 0.30 (b)
			2	5.25 ± 0.20 (b)	4.25 ± 0.04 (b)	4.08 ± 0.28 (b)	2.95 ± 0.66 (a)	5.99 ± 0.12 (b)
		fat	1	$0.57 \pm 0.18 (a)$	1.82 ± 0.31 (b)	5.06 ± 0.16 (d)	3.95 ± 0.17 (c)	2.51 ± 0.31 (b)
			2	$0.67 \pm 0.15 (a)$	2.89 ± 0.02 (b)	$5.81 \pm 0.21 (d)$	4.64 ± 0.23 (c)	1.99 ± 0.06 (b)
t1.8	Antioxidant activity	DPPH ^a	1	$5.1 \pm 1.7 (a)$	$50.2 \pm 3.5 (d)$	$37.4 \pm 3.9 (c)$	40.4 ± 2.3 (c)	$40.0 \pm 2.8 (c)$
			2	$15.1 \pm 1.4 (b)$	$50.3 \pm 6.0 (d)$	$41.8 \pm 1.4 (c)$	$50.7 \pm 3.7 (d)$	$54.5 \pm 0.4 (d)$
		FRAP ^a	1	– (a)	$21.1 \pm 0.8 (d)$	$10.2 \pm 0.7 (b)$	55.0 ± 2.3 (e)	$19.1 \pm 1.1 (c)$
			2	– (a)	$25.8 \pm 1.2 (d)$	11.3 ± 0.3 (b)	49.7 ± 1.6 (e)	$18.8 \pm 0.7 (c)$
		TPC ^b	1	$0.04 \pm 0.02 (a)$	1.69 ± 0.03 (b)	1.68 ± 0.20 (bc)	2.31 ± 0.02 (c)	1.15 ± 0.06 (b)
			2	$0.03 \pm 0.02 (a)$	2.11 ± 0.06 (c)	0.91 ± 0.02 (b)	2.53 ± 0.04 (c)	1.44 ± 0.05 (b)
t1.9	Fatty acids	C10		$5.9 \pm 0.4 (a)$	4.5 ± 0.3 (a)	$4.0 \pm 1.3 (a)$	2.8 ± 0.4 (a)	$3.2 \pm 1.0 (a)$
				$17.6 \pm 3.5 (b)$	$10.4 \pm 2.3(b)$	$7.8 \pm 2.8 (ab)$	18.8 ± 0.2 (b)	$12.9 \pm 1.2 (b)$
		C14		$9.9 \pm 0.4 (ab)$	$10.6 \pm 1.1 (ab)$	12.0 ± 2.5 (b)	13.9 ± 0.9 (b)	$15.5 \pm 0.6 (b)$
				$10.3 \pm 1.2 (ab)$	13.1 ± 0.2 (b)	$10.2 \pm 0.4 (ab)$	$7.5 \pm 0.4 (a)$	11.3 ± 0.3 (b)
		C16		$18.8 \pm 0.5 (c)$	$12.7 \pm 2.8 (ab)$	$13.8 \pm 1.1 (b)$	$12.1 \pm 0.2 (ab)$	$14.4 \pm 1.1 (b)$
				$16.3 \pm 2.0 (c)$	$11.8 \pm 0.9 (a)$	$10.0 \pm 0.4 (a)$	$9.6 \pm 0.2 (a)$	13.6 ± 0.3 (b)
		C18:1		$28.8 \pm 0.8 (b)$	$44.9 \pm 7.5 (c)$	46.0 ± 0.6 (c)	$46.9 \pm 0.3 (c)$	$33.1 \pm 0.7 (b)$
				$16.7 \pm 2.6 (a)$	$46.5 \pm 0.2 (c)$	$46.5 \pm 3.6 (c)$	31.9 ± 2.5 (b)	$33.3 \pm 1.1 (b)$
		C18:2		$4.8 \pm 0.2 (a)$	$7.0 \pm 1.1 (a)$	$12.0 \pm 0.4 (d)$	$10.0 \pm 0.2 (bc)$	$11.7 \pm 0.2 \text{ (cd)}$
				$8.4 \pm 1.1 (ab)$	$9.1 \pm 1.8 (b)$	$11.1 \pm 0.2 (c)$	$7.5 \pm 0.7 (a)$	$8.9 \pm 0.4 (ab)$
		C18:3		2.3 ± 0.2 (b)	$1.4 \pm 0.2 (a)$	3.1 ± 0.2 (b)	3.4 ± 0.2 (b)	$3.8 \pm 0.2 (b)$
				5.4 ± 0.4 (c)	_	2.1 ± 0.6 (b)	_	2.3 ± 0.3 (b)
		C20:5		$5.0 \pm 0.2 (ab)$	5.9 ± 1.2 (ab)	8.3 ± 0.2 (b)	$6.7 \pm 0.2 (ab)$	$6.8 \pm 0.2 (ab)$
				$4.8 \pm 0.2 (ab)$	$5.9 \pm 0.2 (ab)$	$5.8 \pm 0.2 (ab)$	$4.5 \pm 0.2 (a)$	$4.0 \pm 0.2 (a)$
		C22:6		$2.8 \pm 0.1 (a)$	$2.2 \pm 0.2 (a)$	$2.5 \pm 0.2 (a)$	$2.3 \pm 0.2 (a)$	$3.3 \pm 0.2 (a)$
				$7.5 \pm 0.2 (b)$	=	$0.7 \pm 0.2 (a)$	_	2.2 ± 0.3 (a)

a,b,c and d: homogeneous groups obtained from the statistical analysis (ANOVA), for the different species of seaweeds and the different batches used (n = 3). t1.10

mmol equivalents of Trolox/g DW. t1.11

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(mmol equivalents of Gallic Acid/g DW).

the same pattern in the seaweed samples but DPPH values were slightly higher than FRAP values. The DPPH method measures free radicalscavenging ability and higher values might be due to higher levels of phenolic compounds. Catechin, epigallocatechin, phlorotaninns and fucoxantines have all been reported in brown seaweed (Chakraborty, Praveen, Vijayan, & Rao, 2013; Jaime, Pulido, & Saura-Calixto, 2001; Kuda, Tsunekawa, Goto, & Araki, 2005; Langley-Evans, 2000; Meenakshi, Umayaparvathi, Arumugam, & Balasubramanian, 2011). The DPPH data reported here may also indicate the presence of secondary metabolites with antioxidant activity, such as phlorotannins and fucoxanthin, which have previously been reported to be active compounds with antioxidant properties in brown seaweeds (Meenakshi et al., 2011). The antioxidant values exhibited in the present study may be due to the presence of such compounds or any other potential antioxidants with centre/s of unsaturation.

Regarding the FRAP assay, the reducing abilities of chemical compounds, are generally dependent on the presence of reductones, which have been shown to impart antioxidant action by breaking the free radical chain reaction. The presence of antioxidants (reductants) in the samples leads to reduction of the Fe³⁺/ferricyanide complex to its Fe²⁺ form. The results obtained in the present study are in accordance with earlier reports, where it was suggested that brown seaweeds show potential reducing abilities. The reduced form of iron (Fe²⁺) can stimulate and accelerate lipid peroxidation by decomposing lipid hydroperoxides into peroxyl and alkoxyl radicals, that can themselves, abstract hydrogen and perpetuate the chain reaction of lipid peroxidation. As a result, chelators of Fe²⁺ ion can be considered as potential inhibitors of lipid peroxidation. However, the chelating abilities of the samples in the current study may also be due to the presence of different types of polysaccharides. Molecules with hydroxyl, sulfhydryl, carbonyl, and phosphate groups have been reported to possess 345 favourable structure-function configuration resulting in Fe²⁺ chelating 346 abilities. Compounds such as phenolic acids, flavonoid, quercetin, 347 and phenolic glycosides are known to chelate transition metal ions 348 like Fe²⁺ iron. These active compounds might have a synergistic effect, 349 playing an important role in antioxidant activity by the inhibition of ox- 350 idation and chelating effects (Cho, Lee, Kang, Won, & You, 2011; Costa, 351 Gonçalves, Andrade, Valentão, & Romano, 2011; Rajauria, Jaiswal, 352 Abu-Ghannam, & Gupta, 2010).

3.3. Fatty acid composition

The fatty acid composition of the two batches of seaweed samples 355 is given in Table 1. The most abundant fatty acids were oleic acid C_{18:1} 356 (21.9 to 41.45%), myristic $C_{14:0}$ (6.63 to 26.75%) and palmitic $C_{16:0}$ 357 (9.23 to 16.91%) while the results are comparable to those presented 358 by other authors for green and brown seaweeds. Ortiz et al. (2006) re- 359 ported that oleic acid was the most abundant monounsaturated fatty 360 acid in samples of brown seaweeds collected from the coastal area 361 of Northern Chile while, palmitic was found to be the most abundant 362 fatty acid by other authors (16 to 63% of total fatty acids) (Sanchez-363 Machado et al., 2004; Yaich et al., 2011). In the present study, the per- 364 centages of fatty acids differed among the species of seaweeds; Laminar- 365 ia, contained the lowest percentage of myristic (10.1 \pm 0.03%) and oleic 366 $(22.7 \pm 8.6\%)$ but the highest percentage of palmitic $(17.5 \pm 1.8\%)$ 367 contrary to other species such as F. vesiculosus or P. canaliculata which Q17Q18 contained low percentages in palmitic (10.8 \pm 1.6 and 11.3 \pm 1.8% 369 respectively) but higher contents of oleic (39.3 \pm 1.5 and 46.3 \pm 0.4% 370 respectively). Finally, there were no significant differences in the 371 percentages of the long-chain omega-3 fatty acids (EPA: C20:5 372

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eicosapentaenoic acid, and DHA: $C_{22:6}$ docosahexaenoic acid), among the different seaweeds species, although there were seasonal differences in EPA content for *Pelvetia* and *Fucus sp.* Variations in fatty acid contents are attributable to both environmental and genetic differences. Although seaweeds are not a conventional source of energy (their total lipid content is low compared to other foods), their polyunsaturated fatty acid contents can be as high as those of terrestrial vegetables (Sanchez-Machado et al., 2004).

3.4. Free amino acids, nucleotides and umami contribution

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The free amino acid composition (mg/g of dry weight) is illustrated in Table 2. It is important to point out, the high alanine content in the seaweeds collected in July and August of L digitata (4.1 ± 0.2 mg/g of dry weight) compared to those collected earlier for the same species, but also compared to the others. Glutamic acid was particularly high in P. canaliculata and F. spiralis, while aspartic acid was the highest amino acid in F. spiralis. Similar results were found by other authors such as Yaich et al. (2011) and Dawczynski et al. (2007) who found that aspartic acid and glutamic acid constituted, a substantial amount of the total amino acids (26%) for green and brown seaweeds. The contents of glutamic and aspartic acid were of the same order of magnitude as those found for other foods such as tomatoes or potatoes (Coulier et al., 2011; Morris et al., 2007; Oruña-Concha et al., 2007), but in considerably lower amounts than have been found in some species of mushrooms (40 mg/g dry weight) (Beluhan & Ranogajec, 2011).

The nucleotide composition (µg/g of dry weight) for the five seaweeds samples is given in Table 2. These values ranged from 0.20 \pm 0.02 to 364.3 \pm 13.2 µg/g of dry weight, and were of the same order of magnitude as reported in other foods such as tomatoes, potatoes or some varieties of mushrooms (60 to 300 µg/g of dry weight) (Cho, Choi, & Kim, 2010; Morris et al., 2007; Oruña-Concha et al., 2007). Nevertheless, it is important to highlight that the amount of the different nucleotides was found to be ten times higher for *F. vesiculosus*, compared with the other seaweeds, which is similar to the concentrations found by Beluhan and Ranogajec (2011) in some species of mushrooms.

It has previously been suggested that four 5'-nucleotides (5'-AMP, 5'-IMP, 5'-GMP, and 5'-XMP [xanthosine monophosphate]) contribute to umami taste in mushrooms; and the umami taste would synergistically

increase by the combination of umami amino acids and the umami $5'_{1}$ – 410 nucleotides (Yamaguchi, Yoshikawa, Ikeda, & Ninomiya, 1971). The 411 EUC value of 100% indicates that the umami intensity of sample per g 412 of dry matter is equivalent to the umami intensity of 1 g of MSG 413 (monosodium glutamate-like). The EUC values of the different seaweed 414 species are illustrated in Table 2, and they varied widely, ranging from 415 0.31 ± 0.05 in *Laminaria d*. (batch 1) to $74.5 \pm 27.0\%$ in *F. vesiculosus* 416 (batch 2). The high levels of aspartic and glutamic acids, in combination 417 with the nucleotides content might be responsible for the characteristic 418 flavour and taste of seaweeds.

3.5. Volatiles analysis

A total of 23 compounds were detected and identified in the aqueous 421 extracts of the 5 seaweeds. Volatile compounds identified in the dif- 422 ferent seaweed samples are presented in Table 3 and can be classified 423 as aldehydes, alcohols, esters, ketones, acids and aromatic compounds. 424 Five key compounds, (hexanal, heptanal, nonanal, 1-octen-3-ol and 425 2,4-heptadienal), which have previously been described as giving rise 426 to fishy notes (Ganeko et al., 2008; Giri, Osako, & Ohshima, 2010) 427 were studied in more detail. They were quantified using external cali- 428 bration curves and the Friedman test was applied to study any differences in their concentrations between the aqueous seaweed extracts 430 (Fig. 1).

Although Laminaria had the lowest fat content, it contained the 432 highest amount of aldehydes. These volatile compounds can contribute 433 desirable aroma as well as an undesirable rancid odour and flavour during spoilage of fat and fatty foods, due to their low threshold values (Giri et al., 2010). Straight and branched-chain aldehydes generally provide 436 herbaceous, grassy and pungent aromas, while unsaturated aldehydes are linked with vegetable and fishy notes (Giri et al., 2010). The formation of aldehydes, including hexanal, heptanal, octanal and nonanal 439 can also be attributed to the decomposition of lipid hydroperoxides 440 and peroxyl radicals. From all this, it could be suggested that, the aldehydes found in this study such as hexanal, heptanal nonanal and 2,4-42 heptadienal may play a major role in determining the volatiles of the 443 seaweed samples.

Moreover, branched-chain alcohols like 1-octen-3-ol may contrib- 445 ute significantly to the aroma as they are known to have low odour 446

Q7 Table 22.2 Quantities of 5'ribonucleotides, amino acids and Equivalent Umami Concentration found in the different species of seaweeds and the different batches used (n = 3).

t2.3		Batch	Laminaria digitata	Ascophyllum nodosum	Pelvetia canaliculata	Fucus vesiculosus	Fucus spiralis
t2.4	5'Nucleotides ^a						
t2.5	UMP	1	142.1 ± 6.4	97.5 ± 13.7	167.4 ± 17.9	1754.9 ± 119.7	259.0 ± 38.3
		2	81.7 ± 4.7	_	294.7 ± 10.0	1946.9 ± 100.5	104.0 ± 10.0
t2.6	IMP	1	/-	_	_	1229.3 ± 109.5	15.5 ± 0.6
		2				1390.0 ± 87.7	11.3 ± 0.3
t2.7	GMP	1	69.7 ± 26.7	96.2 ± 28.0	87.3 ± 6.9	3873.0 ± 295.0	364.3 ± 13.2
t2.8		2	110.4 ± 0.7	187.5 ± 51.2	$136.4 \pm -$	3908.5 ± 308.9	235.9 ± 10.8
t2.9	AMP	1	_	55.7 ± 4.1	_	74.3 ± 0.2	125.8 ± 9.7
		2		_		_	_
$\begin{array}{c} t2.10 \\ t2.11 \end{array}$	Amino acids ^b						
t2.12	GLU ^c	1	0.15 ± 0.03	0.72 ± 0.16	1.02 ± 0.09	0.43 ± 0.13	1.65 ± 0.13
		2	0.61 ± 0.26	0.47 ± 0.12	1.32 ± 0.25	0.54 ± 0.25	1.25 ± 0.29
t2.13	ASP ^c	1	0.05 ± 0.02	1.06 ± 0.13	0.22 ± 0.02	0.25 ± 0.06	2.75 ± 0.12
		2	0.23 ± 0.06	1.44 ± 0.27	0.21 ± 0.07	0.71 ± 0.08	3.09 ± 0.47
t2.14	Alanine	1	0.72 ± 0.07	0.70 ± 0.02	0.31 ± 0.02	0.35 ± 0.02	2.62 ± 0.09
		2	4.13 ± 0.16	0.39 ± 0.02	1.01 ± 0.02	0.44 ± 0.02	1.37 ± 0.02
t2.15	Proline	1	0.005 ± 0.002	0.011 ± 0.002	0.010 ± 0.002	0.017 ± 0.002	0.058 ± 0.008
		2	0.025 ± 0.003	0.014 ± 0.002	0.017 ± 0.002	0.023 ± 0.002	0.040 ± 0.003
t2.16	Asparagine	1	_	0.154 ± 0.019	0.075 ± 0.013	0.483 ± 0.005	0.230 ± 0.004
		2		0.069 ± 0.002	0.051 ± 0.006	0.152 ± 0.018	0.274 ± 0.046
t2.17	EUC ^d	1	0.31 ± 0.05	2.29 ± 0.06	1.75 ± 0.31	55.44 ± 12.61	21.05 ± 6.41
		2	1.81 ± 0.48	3.03 ± 0.09	3.04 ± 0.41	74.44 ± 27.01	13.83 ± 2.76

a μg/g of dry weight.

t2.18

t2.19

t2.20

t2.21

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b mg/g of dry weight.

^c Umami amino acids (glutamic acid and aspartic acid).

d g MSG/100 g.

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t3.2 t3.3

t3.36

t3.37t3.38

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Table 3 Retention time, retention index and odour descriptors of volatile compounds found in the different species of seaweeds and the different batches used (n = 3).

t3.3		RT	RI	Identification	Odour description
t3.4	Aldehydes				
t3.5	Hexanal	16.551	1080	MS, RI Std	Fishy, grass ^{a-c}
t3.6	Heptanal	21.604	1170	MS, RI Std	Dry fish ^{a,d} , citrus fruit ^{b-d} , green, fatty, pesticide, solvent, smoky, rancid, fruity ^d
t3.7	Octanal	26.057	1286	MS, RI Std	Fatty, pungent ^a , fatty-orange odour ^{b.c} , lemon, stew-like, rancid, soapy, citrus, green, flower, fruit, orange ^d
t3.8	2-Heptenal	27.426	1326	MS, RI Std	Pungent green, somewhat fatty aroma ^c
t3.9	Nonanal	30.022	1404	MS, RI Std	Green, fatty ^{a-d} , floral, waxy, sweet, melon, soapy, lavender, citrus fruit ^d
t3.10	2-Octenal	31.328	1512	MS, RI Std	Aromatic, oxidized oil-like ^b , fatty, nutty, burdock-like, sweet, sour, waxy, green, burnt, mushroom ^d
t3.11	2,4-Heptadienal	32.529	1531	MS, RI Std	Fatty, fishy ^{a,c} , aromatic, oxidized oil-like ^b
t3.12	•				
t3.13	Alcohols				
t3.14	1-Penten-3-ol	20.321	1148	MS, RI Std	Burnt, meaty ^a , paint like chemical like ^b grassy-green ^c
t3.15	1-Octen-3-ol	31.795	1520	MS, RI Std	Fishy, grassy ^a , sweet earthy ^c
t3.16	2-ETHYL-1-hexanol	33.142	1541	MS, RI	Green rose ^a
t3.17	4-Hepten-1-ol	33.596	1549	MS, RI Std	Fishy ^c
t3.18					
t3.19	Esters				
t3.20	Ethyl acetate	7.623	692	MS, RI	Fruity orange a,d , acetic, ethereal odour c , caramel, sweet, solvent-like, acid, buttery, pungent, orange d
t3.21					
t3.22	Ketones				
t3.23	4-Methyl-2- Heptanone	22.534	1187	MS, RI	ND
t3.24	1-Octen-3-one	26.532	1301	MS, RI Std	Mushroom like ^{b,c} , metallic, dirty, dust, herb ^d
t3.25	6-Methyl-5-hepten-2-one	27.927	1341	MS, RI Std	Sweet, fruity a,c,d , fatty c , mushroom, earthy, vinyl, rubber, woody, blackcurrant, boiled fruity d
t3.26					
t3.27	Acids				
t3.28	Acetic acid	32.154	1525	MS, RI	Pungent odour ^{c,d} , sour, vinegar ^d
t3.29	4-Hydroxy butanoic acid	37.994	1642	MS, RI	ND
t3.30	2-Ethyl hexanoic acid	46.424	1900	MS, RI	ND
t3.31					
t3.32	Aromatic compounds				
t3.33	Methylene chloride	9.131	927	MS, RI	Chloroform-like odour ^d
t3.34	Benzaldehyde	33.014	1539	MS, RI	Bitter almond ^{a,c,d} , burnt sugar, woody ^d
t3.35	Phenol	34.589	1565	MS, RI	Herbal, anisic ^a sweet, tarry odour ^c , medicinal odour ^d

^a Giri et al. (2010).

threshold values. They can be mostly produced by secondary decomposition of hydroxyperoxides of fatty acids, but some of them might also come from carbohydrates by the glycolysis and/or from amino acids via the Ehrlich pathway (Giri et al., 2010). As expected, there were sig- 450 nificant differences in the volatile composition between samples, where 451 their overall aroma was enhanced by the presence of aldehydes and 452

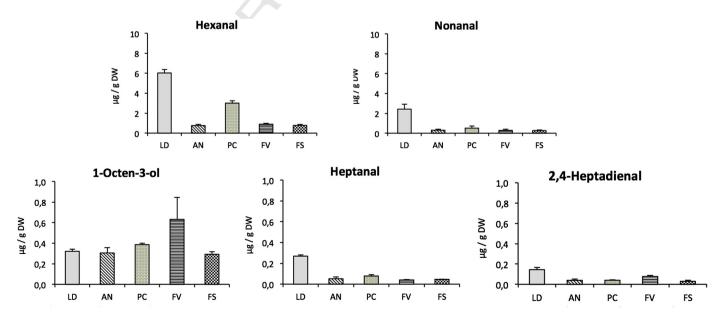


Fig. 1. Concentration of the most relevant seafood volatile compounds in the aqueous extracts used for sensory evaluation (µg/g DM), quantified using external calibration curves (LD: Laminaria digitata, AN: Ascophyllum nodosum, PC: Pelvetia canaliculata, FV: Fucus vesiculosus, and FS: Fucus spiralis).

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Ganeko et al. (2008).

fao.org.

^d pherobase.org.

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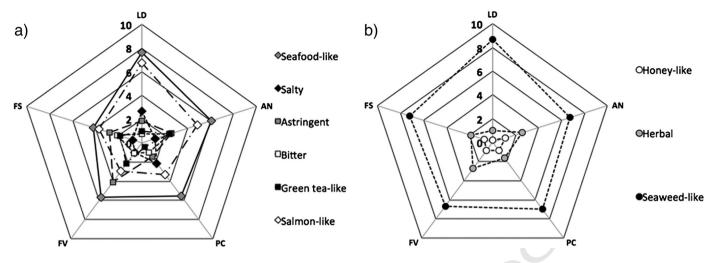


Fig. 2. Spider diagram obtained for the different attributes of the different seaweed aqueous extracts. (LD: Laminaria digitata, AN: Ascophyllum nodosum, PC: Pelvetia canaliculata, FV: Fucus vesiculosus, and FS: Fucus spiralis).

alcohols. These compounds have also been found in the volatiles profile of cooked fish or meals containing seafood (Ganeko et al., 2008; Giri et al., 2010).

3.6. Sensory evaluation

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Fig. 2 shows the spider diagram obtained for the different attributes studied for the aqueous seaweed extracts. Seaweed-like aroma, seafood-like taste and salmon-like taste, where in general, the attributes with the higher scores which could be expected as those were the attributes more related to "seafood-like". The Friedman test illustrated that panellists were only able to notice significant differences between samples in 3 out of the 9 attributes evaluated. In fact, only Laminaria sp. extract was significantly different from all the others in terms of aroma, being the one with the strongest seaweed-like aroma, and the mildest honey-like aroma; and showing the strongest seafood-like taste. Despite the fact that Laminaria showed the highest score for saltiness, as could be expected due to its high concentration in NaCl compared to the other seaweeds, the difference was not significant. The results suggest that the panellists did not associate umami taste with seafood taste or seaweed aroma, as Laminaria had the lowest EUC (Table 2). This could be due to the assessors used were untrained subjects unfamiliar with the characteristics of the typical umami taste, however, this type of panel has previously been used for that kind of

assessment and though the performance of the untrained panels 475 would not be as good as if they had been trained, they were able to dis-476 tinguish between samples (Clapperton & Piggott, 1979; Husson' & 477 Pagés, 2003). Therefore its sensory attributes could be mainly due to 478 its high salt content together with high levels of the volatile compounds, 479 hexanal, heptanal, nonanal and 2,4-heptadienal.

Fig. 3 illustrates the PCA conducted to simplify the interpretation of 482 the relationships between the seaweed samples and their chemical, 483 volatile and sensory profile. The first three components explain 94% of 484 the total variance. First principal component (PC1, 54%) separated 485 *Laminaria* from the other samples, which presented the lower antioxi-486 dant activity, highest levels of aldehydes and highest scores for 487 seaweed-like odour and seafood-like taste. The second principal com-488 ponent (PC2, 23%) differentiated *F. vesiculosus* from the other samples. 489 *Fucus v* possessed the highest nucleotide values as well as the highest 490 concentration of 1-octen-3-ol. Finally, the third principal component 491 (PC3, 17%) differentiated *Fucus s*. from the other seaweed samples 492 mostly in terms of the amino acid content. As suggested above, the differences in concentrations of the various compounds, such as the high 494 contents of aldehydes and salt in *Laminaria*, or the high content of 495

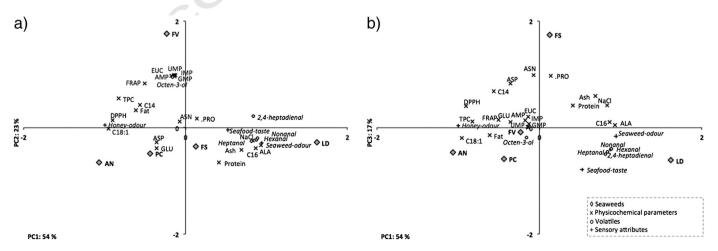


Fig. 3. Biplots for the different seaweeds (LD: Laminaria digitata, AN: Ascophyllum nodosum, PC: Pelvetia canaliculata, FV: Fucus vesiculosus, and FS: Fucus spiralis), depending on their composition: chemical values (ash, NaCl, protein and fat content; antioxidant activity (DPPH, FRAP and TFC), fatty acids and amino acid composition as well as volatiles and sensory attributes. (PC1: 54%, PC2: 23% and PC3: 17%) obtained by means of the PCA.

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alcohols (1-octen-3-ol) and nucleotides of F. vesiculosus, would be responsible for the different sensory profiles obtained by the panellists.

4. Conclusions

The chemical composition of the five brown edible seaweeds object of this study was in general terms comparable, with the composition of other brown seaweeds harvested in other areas such as the coast of Spain, Chile, or Norway among others.

The sensory differences observed between the five samples investigated must be attributed to their different chemical compositions. Laminaria and F. vesiculosus differ significantly from each other and the other species both in terms of their volatiles and sensory profiles, as well as their chemical composition.

F. vesiculosus presented high lipid content as well as high level of nucleotides, while Laminaria had the lowest lipid and highest salt contents. The fatty acids profile of the samples was dominated by oleic acid, followed by myristic and palmitic acids, although the amounts of them varied between the different seaweeds. The high concentration of nucleotides together with the high amounts of aspartic and glutamic acids may influence the characteristic flavour and taste of *F. vesiculosus*.

The high antioxidant activity of the seaweed extracts indicated they could potentially be used as flavour stabilisers specially Fucus sp. and A. nodosum.

Volatiles analysis emphasised the differences between Laminaria and F. vesiculosus compared to the other species. Besides having the lowest lipid content, Laminaria happened to be the seaweed with the highest concentration of lipid-derived aldehydes, and that might be the reason why it resented intense honey-like and seaweed-like odour, as well as an intense seafood-like taste.

The importance of these results is the possibility of using locally harvested brown seaweeds, especially Laminaria and F. vesiculosus which due to their sensory, volatile and chemical composition, could be used to enhance the characteristic umami taste of some foods and/or reduce the need for added salt, as well as providing components possessing antioxidant activity.

5. Uncited references

http://www.pherobase.com, n.d www.netalgae.eu, n.d

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References

- AOAC (2000). Association of official analytical chemists. Official methods of analysis (Washington, DC, USA)
- Beluhan, S., & Ranogajec, A. (2011). Chemical composition and non-volatile components of Croatian wild edible mushrooms, Food chemistry, 124(3), 1076-1082
- Benzie, I. F., & Szeto, Y. T. (1999). Total antioxidant capacity of teas by the ferric reducing/ antioxidant power assay. Journal of Agricultural and Food Chemistry, 47(2), 633-636. Brand-Williams, W., Cuvelier, M. E., & Berset, C. (1995), Use of a free radical method to
- evaluate antioxidant activity. LWT food science and technology, 28(1), 25-30. Bub, A., Watzl, B., Abrahamse, L., Delince, H., Adam, S., Wever, J., et al. (2000). Moderate intervention with carotenoid-rich vegetable products reduces lipid peroxidation in

men 1. Human nutrition and metabolism, 2200-2206.

- Chakraborty, K., Praveen, N. K., Vijayan, K. K., & Rao, G. S. (2013). Evaluation of phenolic contents and antioxidant activities of brown seaweeds belonging to Turbinaria spp. (Phaeophyta, Sargassaceae) collected from Gulf of Mannar, Asian Pacific Journal of Tropical Biomedicine, 3(1), 8-16.
- Chandrasekar, D., Madhusudhana, K., Ramakrishna, S., & Diwan, P. V. (2006), Determination of DPPH free radical scavenging activity by reversed-phase HPLC: a sensitive screening method for polyherbal formulations. Journal of Pharmaceutical and Biomedical Analysis, 40(2), 460-464.

- Cho, I. H., Choi, H. K., & Kim, Y. S. (2010). Comparison of umami-taste active components 557 in the pileus and stipe of pine-mushrooms (Tricholoma matsutake Sing.) of different 558 grades, Food Chemistry, 118(3), 804-807.
- Cho, M., Lee, H. -S., Kang, I. -J., Won, M. -H., & You, S. (2011). Antioxidant properties of 560 extract and fractions from Enteromorpha prolifera, a type of green seaweed. Food 561 Chemistry, 127(3), 999-1006.
- Clapperton, J. F., & Piggott, J. R. (1979). Flavour characterization by trained and untrained 563 assessors. Journal of the Institute of Brewing, 85(5), 275-277. 564

562

581

583

592

593

- Costa, P., Gonçalves, S., Andrade, P. B., Valentão, P., & Romano, A. (2011). Inhibitory effect 565 of Lavandula viridis on Fe2+-induced lipid peroxidation, antioxidant and anti-566 cholinesterase properties. Food Chemistry, 126(4), 1779–1786.
- Coulier, L., Bas, R., Hekman, M., Van der Werff, B. J. C., Burgering, M., & Thissen, U. (2011). 568 Comprehensive analysis of umami compounds by ion-pair liquid chromatography 569 coupled to mass spectrometry. Journal of Food Science, 76(7), C1081-C1087. 570
- Dawczynski, C., Schubert, R., & Jahreis, G. (2007). Amino acids, fatty acids, and dietary 571 fibre in edible seaweed products. Food Chemistry, 103(3), 891-899. 572
- Delahunty, C., Mc Cord, F., O'Neille, E., & Morissey, P. (1997). Sensory characterization of 573 cooked hams by untrained consumers using free-choice profiling. Food Quality and 574 Preference, 8, 384-388.
- Díaz-Rubio, M. E., Pérez-Jiménez, J., & Saura-Calixto, F. (2009). Dietary fiber and antioxi-576 dant capacity in Fucus vesiculosus products. International Journal of Food Sciences 577 and Nutrition, 60(s2), 23-34. 578
- Elmore, J. S., Koutsidis, G., Dodson, A. T., Mottram, D. S., & Wedzicha, B. L. (2005). Mea-579 surement of acrylamide and its precursors in potato, wheat, and rye model systems. 580 Journal of Agricultural and Food Chemistry, 53(4), 1286–1293. 582
- Fleurence, J. (1999). Seaweed proteins: biochemical, nutritional aspects and potential uses. Trends in Food Science & Technology, 10, 25-28.
- Fleurence, J. (2004). Seaweed proteins. In R. Y. Yada (Ed.), Proteins in food processing 584 (pp. 197-213). Cambridge: Woodhead Publishing Limited. 585
- Fleurence, J., Gutbier, G., Mabeau, S., & Leray, C. (1994). Fatty acids from 11 marine macroalgae of the French Brittany coast. Journal of Applied Phycology, 6, 527-532.
- Fleurence, J., Morançais, M., Dumay, J., Decottignies, P., Turpin, V., Munier, M., et al. (2012). What are the prospects for using seaweed in human nutrition and for marine animals raised through aquaculture? Trends in Food Science & Technology, 27, 57-61.
- Ganeko, N., Shoda, M., Hirohara, I., Bhadra, A., Ishida, T., Mathsuda, H., et al. (2008). Analysis of volatile flavor compounds of sardine (Sardinops melanostica) by solid phase microextraction. Journal of Food Science, 73(1), S83-S88.
- Giri, A., Osako, K., & Ohshima, T. (2010). Identification and characterisation of headspace volatiles of fish miso, a Japanese fish meat based fermented paste, with special emphasis on effect of fish species and meat washing. Food Chemistry, 120(2), 621-631.
- Gómez-Ordóñez, E., Jiménez-Escrig, A., & Rupérez, P. (2010). Dietary fibre and physicochemical properties of several edible seaweeds from the northwestern Spanish coast. Food Research International, 43(9), 2289-2294.
- Gupta, S., & Abu-Ghannam, N. (2011a). Recent developments in the application of seaweeds or seaweed extracts as a means for enhancing the safety and quality attributes 601 of foods. Innovative Food Science and Emerging Technologies, 12, 600-609.
- Gupta, S., & Abu-Ghannam, N. (2011b). Bioactive potential and possible health effects of 603 edible brown seaweeds. Trends in Food Science & Technology, 22(6), 315-326. http://www.pherobase.com (Accessed 20.07.2013).
- Husson, F., Le Dien, S., & Pagés, J. (2001). Which value can be granted to sensory profiles given by consumers. Food Quality and Preference, 12, 291-296.
- Husson', F., & Pagés, J. (2003). Compartion of sensory profiles done by trained and untrained juries: methodology and results. Journal of Sensory Studies, 18, 453-464.
- Ito, K., & Kanji, H. (1989). Seaweed chemical composition and potential food uses. Food reviews international, 5(1), 101-144. IUPAC (1992). http://lipidlibrary.aocs.org/topics/ester_93/file.pdf (Accessed 20.07.2012).
- Jaime, I., Pulido, R., & Saura-Calixto, F. (2001). Antioxidant activity of fresh and processed edible seaweeds534, 530-534.
- Kuda, T., Tsunekawa, M., Goto, H., & Araki, Y. (2005). Antioxidant properties of four edible 615 algae harvested in the Noto Peninsula, Japan. Journal of Food Composition and Analysis, 18(7), 625-633.
- Magalhães, L. M., Santos, F., Segundo, M. A., Reis, S., & Lima, J. L. F. C. (2010). Rapid micro-618 plate high-throughput methodology for assessment of Folin-Ciocalteu reducing ca- 619 pacity. Talanta, 83(2), 441-447.
- Meenakshi, S., Umayaparvathi, S., Arumugam, M., & Balasubramanian, T. (2011). In vitro 621 antioxidant properties and FTIR analysis of two seaweeds of Gulf of Mannar. Asian Pacific Journal of Tropical Biomedicine, 1(1), S66-S70.
- Mišurcová, L. (2011). Chemical composition of seaweeds. In S. K. Kim (Ed.), Handbook of 624 Marine Macroalgae: Biotechnology and Applied Phycology (pp. 173-192). Chichester: 625 Iohn Wiley & Sons Ltd. 626
- Mišurcová, L., Ambrožová, J., & Samek, D. (2011). In S. K. Kim (Ed.), Seaweed lipids 627 as nutraceuticals. Advances in food and nutrition research, Vol. 64. (pp. 339–355). Burlington: Academic Press, Elsevier Inc. 978-0-12-387669-0.
- Mišurcová, L., Machů, L., & Orsavová, I. (2011), In S. K. Kim (Ed.), Segweed minerals as 630 nutraceuticals. Advances in food and nutrition research, Vol. 64. (pp. 371-390). Burlington: 631 Academic Press, Elsevier Inc978-0-12-387669-0.
- Morris, W. L., Ross, H., Ducreux, L., Bradshaw, I., Brvan, G., & Taylor, M. (2007), Umami 633 compounds are a determinant of the flavor of potato (Solanum tuberosum L.). 634 Journal of Agricultural and Food Chemistry, 55, 9627-9633. 635
- Ortiz, J., Bozzo, C., Navarrete, E., Osorio, A., & Rios, A. (2006). Dietary fiber, amino acid. 636 fatty acid and tocopherol contents of the edible seaweeds Ulva lactuca and Durvillaea 637 antarctica. Food Chemistry, 99, 98-104. 638
- Oruña-Concha, M. J., Methven, L., Blumenthal, H., Young, C., & Mottram, D. (2007). Differ- 639 ences in glutamic acid and 5'-ribonucleotide contents between flesh and pulp of tomatoes and the relationship with Umami Taste. Journal of Agricultural and Food 641 Chemistry, 55, 5776-5780. 642

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Rajauria, G., Jaiswal, A. K., Abu-Ghannam, N., & Gupta, S. (2010). Effect of hydrothermal	www.netalgae.eu (accessed 20.07.2012).	658
processing on colour, antioxidant and free radical scavenging capacities of edible	Xu, X. L., Fan, X., Song, F. H., Zhao, J. L., Han, L. J., Yang, Y. C., et al. (2004). Bromophenols	659
Irish brown seaweeds. International Journal of Food Science & Technology, 45(12),	from the brown alga Leathesia nana. Journal of Asian Natural Products Research, 6,	660
2485–2493.	217–221.	661
Rioux, LE., Turgeon, S. L., & Beaulieu, M. (2009). Effect of season on the composition	Xu, X., Song, F., Wang, S., Li, S., Xiao, F., Zhao, J., et al. (2004). Dibenzyl bromophenols with	Q2:
of bioactive polysaccharides from the brown seaweed Saccharina longicruris.	diverse dimerization patterns from the brown alga Leathesia nana. Journal of Natural	663
Phytochemistry, 70(8), 1069–1075.	Products, 67, 1661e1666.	664
Sanchez-Machado, D. I., Lopez-Cervantes, J., & Lopez-Hernandez, P. L. (2004). Fatty acids,	Yaich, H., Garna, H., Besbes, S., Paquot, M., Blecker, C., & Attia, H. (2011). Chemical compo-	665
total lipid, protein and ash contents of processed edible seaweeds. Food chemistry, 85,	sition and functional properties of Ulva lactuca seaweed collected in Tunisia. Food	666
439–444.	Chemistry, 128(4), 895–901.	667
Solms, J., & Wyler, R. (1979). Taste components of potatoes. In J. C. Boudreau (Ed.), Food	Yamaguchi, S., Yoshikawa, T., Ikeda, S., & Ninomiya, T. (1971). Measurement of the rela-	668
taste chemistry (pp. 175–184). Washington, DC: ACS Symposium,	tive taste intensity of some L- α -amino acids and 5'-nucleotides. Journal of Food	669
Wang, T., Jónsdóttir, R., & Ólafsdóttir, G. (2009). Total phenolic compounds, radical scav-	Science, 36, 846–849.	670
enging and metal chelation of extracts from Icelandic seaweeds. Food chemistry,	Yang, J. H., Lin, H. C., & Mau, J. L. (2001). Non-volatile taste components of several com-	671
116(1), 240–248.	mercial mushrooms. Food Chemistry, 72(4), 465–471.	672